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Solution Quiz 10 Chemical Engineering Thermodynamics 2020 Chemical
Engineering Thermodynamics April 2, 2020 The K-ratio is used to calculate the
equilibrium distribution of a component in a mixture between vapor and liquid
phases We went through five methods to determine the K-ratio For n-octane at
100°C at 01 MPa calculate the K-ratio: a) Using Raoult's Law and P sat from: -The
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are also seminars (20 hours), conceived as integrated, highly interactive sessions,
in which larger problems are discussed from different viewpoints and solved by
various methods. These sessions aim at a deeper insight into the theory in the

lectures and at its application to practical problems in chemical engineering. Chemical Engineering Thermodynamics - KU Leuven Many of the definitions below are also used in the thermodynamics of chemical reactions.

General basic quantities. Quantity (Common Name/s) (Common) Symbol/s SI Units Dimension

Number of molecules N : dimensionless dimensionless Number of moles n : mol ... $C V = \partial / \partial J K^{-1} [M][L ...$ Table of thermodynamic equations -

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