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easily because: 1) It is simple and

easier to set up. 2) It does not

require constant attention. 3)

Equilibrium is quickly attained,

hence it is not so time-consuming.

4) A small quantity of solution is

needed. Class 12 Chemistry

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Chemistry Quick Revision Notes

Chapter 3 Electrochemistry

Oxidation: It is defined as a loss of electrons while reduction is defined as a gain of electrons. In a redox

reaction, both oxidation and

reduction reaction takes place

simultaneously. Electrochemistry

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Kinetics Rate of a reaction (Average
and instantaneous), factors

affecting rate of reaction:

concentration, temperature,

catalyst; order and molecularity of a

reaction, rate law and specific rate

constant, integrated rate equations

and half life (only for zero and first

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of two or more substances in same or different physical phases. The substances forming the solution are called components of the solution. On the basis of number of components a solution of two components is called binary solution. Chemistry Notes for class 12 Chapter 2 Solutions Class 12 Chemistry Notes PDF Free Download. CBSE Schools Educational Study Material. Class 12 Chemistry Notes PDF - NCERT Solutions Class 12 Chemistry notes on chapter 7 The p-Block Elements are also available for download in CBSE Guide website. CBSE Guide the p-Block Elements class 12 Notes Chemistry CBSE guide notes are the comprehensive notes which covers the latest syllabus of CBSE and NCERT. It includes all the topics

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given in NCERT class 12 Chemistry text book. The p-Block Elements class 12 Notes Chemistry ... P - Block Elements Class 12 Notes Class 12 Chemistry Chapter 7 P - Block Elements Notes- Pdf Download P-Block elements are seen on the right side of the periodic table and consist of six columns. They are so called because in these the valence electron which is commonly known as outermost electron enters the p-orbital.

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